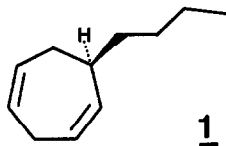


ENANTIOSELECTIVE SYNTHESIS OF DICTYOPTERENE C  
6R(-)-BUTYL-2,5-CYCLOHEPTADIENE  
THE PHEROMONE OF SEVERAL DICTYOTALES (PHAEOPHYCEAE)

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**Abstract:** R(-)-Dictyopterene C (**1**) is a widespread constituent of many marine brown algae (Phaeophyceae). A highly enantioselective synthesis of **1** and its enantiomer ent-**1** via chromatographic separation of the diastereomeric  $\gamma$ -hydroxyphenylethylamide intermediates **3a** and **3b** is described.

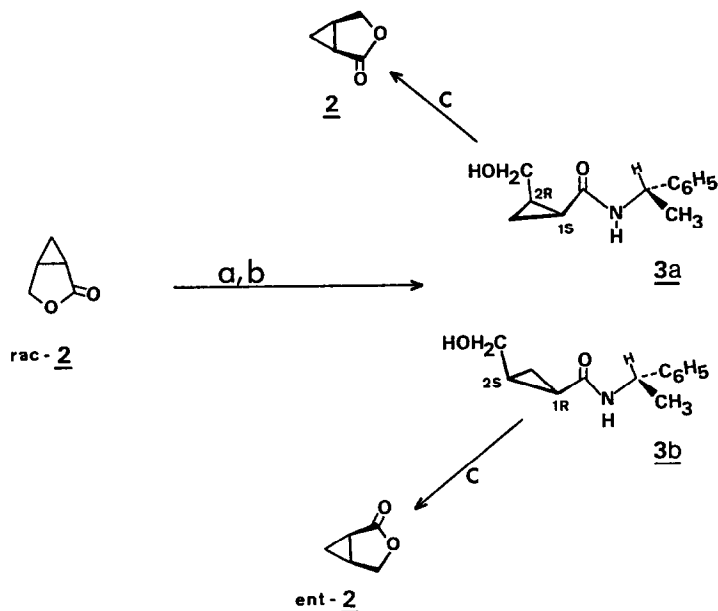


(R)-(-)-Dictyopterene C **1** and several other unsaturated C<sub>11</sub>-hydrocarbons were first isolated by R.E. Moore et al. from the Pacific seaweeds *Dictyopteris plagiogramma* and *D. australis* collected around Hawaii<sup>1,2)</sup>. Its occurrence in several North-Pacific representatives of the same genus was also reported by Kajiwara et al.<sup>3)</sup>. The absolute configuration of **1** was established as (6R)- by ozonolytic degradation of R(-)-**1** to S(-)-butylsuccinic acid<sup>2)</sup>. R(-)-**1** was later identified by Müller et al. as the sperm attractant of female gametes of the North Atlantic phaeophyte *Dictyota dichotoma*<sup>4)</sup>. Very recently we found that dictyopterene C is also the major product of the egg secretions of the Australian phaeophyte *Dictyota prolifera*<sup>5)</sup>. The very small amount of isolated material (ca. 5  $\mu$ g) did not allow for the determination of its absolute configuration. Hence, we approached this question by synthesizing **1** and ent-**1** and to eventually compare their pheromone activity for male gametes of *Dictyota prolifera*.

A promising route to R(-)-**1** of very high optical purity is the chromatographic separation of suitable  $\gamma$ -hydroxyphenylethylamide intermediates according to the method of Helmchen<sup>6)</sup>, which also allows the inherent control of the optical purity of the products obtained<sup>7)</sup>.

Synthesis was started with the readily available lactone<sup>8)</sup> rac-**2** which forms the two diastereomeric amides **3a** and **3b** in 80% yield, if reacted with 2 eq. S(-)-phenylethylamine and 1 eq. 2-hydroxypyridine<sup>6)</sup> as depicted in Scheme 1. Both diastereomers are easily separated by MPLC on silica gel (ca. 400 g silica gel loaded with the amide mixture in 1 g aliquots and eluted with hexane/AcOEt 1:9/v:v;  $\alpha = 1.45$ ).

Scheme 1



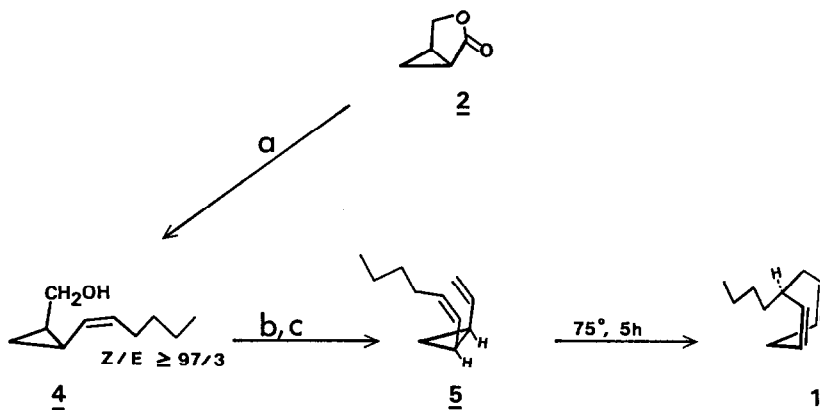
- a) 2 eq. S-(-)-phenylethylamine and 1.1 eq. 2-hydroxypyridine  
 b) separation, see text  
 c) 2 N H<sub>2</sub>SO<sub>4</sub> in dioxane/water (1:1), 12h, 90°

The "anti" orientation of the aromatic ring and the cyclopropyl moiety in (-)-(1S,2R)-**3a**, diminishes its diastereoselective interaction with the stationary phase; therefore this amide elutes first<sup>7,10</sup>) [ $\alpha$ ]<sub>D</sub><sup>25</sup> = -151.4 (c = 3.7, MeOH; 83% yield), clearly separated from (-)-(1R,2S)-**3b**<sup>10</sup>) [ $\alpha$ ]<sub>D</sub><sup>25</sup> = -153.3 (c = 3.98, MeOH; 55% yield). After recrystallization from AcOEt both compounds were obtained diastereomerically pure<sup>9</sup>) as confirmed by HPLC on an analytical column. Support for this configurational assignment comes also from the known absolute configurations of the products **2**<sup>11</sup>) and **1**<sup>2</sup>).

Mild acid hydrolysis<sup>6</sup>) (2N H<sub>2</sub>SO<sub>4</sub> in dioxane/water (1/1; v/v), 16 h, 90°) of **3a** or **3b** furnished the two key intermediates **2** [ $\alpha$ ]<sub>D</sub><sup>25</sup> = -69.5 (c = 6.8, CHCl<sub>3</sub>), [ $\alpha$ ]<sub>D</sub><sup>25</sup> = -61.8 (c = 6, CHCl<sub>3</sub>)<sup>11</sup>) and ent-**2** [ $\alpha$ ]<sub>D</sub><sup>25</sup> = +68.7 (c = 4.6, CHCl<sub>3</sub>); both in 97% yield.

From the two optically pure lactones both enantiomers of dictyopterene C, **1** and ent-**1**, respectively, are readily synthesized as outlined in Scheme 2. Reductive olefination<sup>12</sup>) of **2a** with 1.1 eq. diisobutylaluminium hydride in toluene at -78°, followed by immediate addition of a "salt free" preparation of Ph<sub>3</sub>P=CH(CH<sub>2</sub>)<sub>3</sub>CH<sub>3</sub> in THF according to Bestmann et al.<sup>13</sup>) gives the alcohol **4**,

Scheme 2



- a) 1.1 eq. DIBAH at  $-78^{\circ}$  in toluene, followed by addition of 1.5 eq. "salt free"  $\text{CH}_3(\text{CH}_2)_3\text{CH}=\text{P}(\text{C}_6\text{H}_5)_3$ , then warm up to r.t., cf. ref. 12.  
 b) Pyridiniumchlorochromate  
 c)  $\text{CH}_2=\text{P}(\text{C}_6\text{H}_5)_3$  in THF

$[\alpha]_{\text{D}}^{25} = -81.2$  ( $c = 2.6$ ,  $\text{CH}_2\text{Cl}_2$ ) in 45% yield and -97% stereochemical homogeneity<sup>10,14</sup>). By subsequent oxidation of 4 with pyridiniumchlorochromate and Wittig reaction of the intermediary aldehyde with  $\text{Ph}_3\text{P}=\text{CH}_2$  the already known dictyopterene A isomer<sup>15</sup> 5,  $[\alpha]_{\text{D}}^{25} = -117.6$  ( $c = 2.4$ ,  $\text{CCl}_4$ ) is obtained in 53% overall yield. Upon heating to  $75^{\circ}$  for five hours in a sealed vial 5 rearranges in almost quantitative yield to dictyopterene C 1  $[\alpha]_{\text{D}}^{25} = -15.5$  ( $c = 1.74$ ,  $\text{CHCl}_3$ ),  $[\alpha]_{\text{D}}^{22} = -12$  ( $c = 7.32$ ,  $\text{CHCl}_3$ )<sup>2</sup>). Since this sigmatropic Cope rearrangement is known to proceed stereospecifically over a cis-endo transition state<sup>16</sup>, the optical purity of synthetic 1 only depends on the stereochemical purity of the alcohol 4 which was found to be ~97% Z; accordingly also 1 will have 97% enantiomer excess. Hence, the optical purity of the dictyopterene C isolated by Moore et al.<sup>2</sup>) was not more than ca. 75%.

By the same route, ent-2 was transformed into ent-1  $[\alpha]_{\text{D}}^{25} = +15.1$  ( $c = 0.69$ ,  $\text{CHCl}_3$ ) via the alcohol ent-4  $[\alpha]_{\text{D}}^{25} = +80.0$  ( $c = 2.2$ ,  $\text{CH}_2\text{Cl}_2$ ) and the dictyopterene A isomer ent-5  $[\alpha]_{\text{D}}^{25} = +115.3$  ( $c = 0.89$ ,  $\text{CCl}_4$ ).

Comparative biological activity tests on male gametes of *Dictyota proliferata* with 1 and ent-1 are currently under investigation. Interestingly, male gametes of *Ectocarpus siliculosus*, which respond enantioselectively to the structurally related pheromone ectocarpene (6S-(+)-(1Z)-butenyl-2,5-cycloheptadiene)<sup>17</sup> are unable to distinguish between 1 and ent-1. Their receptor obviously directionally recognizes the  $\pi$ -electron system of the extra double bond within the butenyl-substituent.

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8. Racemic lactone 2 is most conveniently prepared by  $\text{NaBH}_4$ -reduction of the corresponding cyclopropyldicarboxylic acid anhydride.
9. Enantiomerically pure S-(-)-phenylethylamine  $[\alpha]_D^{25} = -40.34$  (neat; e.e. = 99.95%), cf. ref. 6, was used.
10. Spectroscopic data of new compounds: 3a:  $^1\text{H}$  NMR ( $\text{CCl}_4$ ): 0.68-1.73 (m, 4H, cyclopropyl, sek. and tert.-CH); 1.40 (d, CH<sub>3</sub>, J=8.3); 3.60-4.05 (ABX-syst., CH<sub>2</sub>OH: J<sub>AX</sub>=6.7, J<sub>BX</sub>=7.3, J<sub>AB</sub>=13.3); 4.03 (s, OH); 5.03 (qi., CH(CH<sub>3</sub>)C<sub>6</sub>H<sub>5</sub>); 7.28 (s, C<sub>6</sub>H<sub>5</sub>); 7.20 (d, -NH-). - 3b: 0.78-1.72 (m, 4H, cyclopropyl, sek. and tert.-CH); 1.43 (d, CH<sub>3</sub>, J=7.3); 3.40 (s, OH); 3.43-3.98 (ABX-syst., CH<sub>2</sub>OH: J<sub>AX</sub>=5.0, J<sub>BX</sub>=8.3, J<sub>AB</sub>=13.3); 5.70 (qi., CH(CH<sub>3</sub>)C<sub>6</sub>H<sub>5</sub>); 6.75 (d, NH); 7.30 (s, C<sub>6</sub>H<sub>5</sub>). - 4: 0.31 (qu., 1H, sek.-cyclopropyl); 0.93 (t, CH<sub>3</sub>); 0.70-1.10 (m, 1H, sek.-cyclopropyl); 1.10-1.80 (m, 6H, CH<sub>2</sub> and tert.-cyclopropyl); 1.85-2.35 (m, CH<sub>2</sub>CH=CH); 2.45 (s, broad, OH); 3.22-3.75 (m, CH<sub>2</sub>OH); 4.90-5.70 (m, CH=CH). - MS (70 eV): 3a or 3b: 219 M<sup>+</sup>(3), 204(1), 188(21), 163(10), 120(56), 106(51), 105(100), 79(26), 77(38), 59(18), 55(47), 43(27). - 4: 154 M<sup>+</sup>(2), 136(2), 123(3), 110(5), 107(4), 93(13), 81(50), 79(43), 77(18), 67(100), 55(32), 54(33), 41(67), 39(31). All new compounds gave satisfactory microanalyses.
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14. Alcohol 4 is homogeneous to GC. The only evidence for a small trace (ca. 3%) of an E-contaminant was the  $^1\text{H}$  NMR of 5.
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